Dialkylaluminum Hydride-Promoted Cyclodimerization of Silylated 1,3-Enynes via Skeletal Rearrangement

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The dialkylaluminum hydride-promoted reaction of 1-silylalk-3-en-1-ynes gave symmetrical 1,2,3,5-tetrasubstituted benzenes as single regioisomers. The novel cyclodimerization via skeletal rearrangement can be rationalized by an unprecedented mechanism involving sequential hydroalumination, alkene isomerization, carboalumination, carbon-carbon bond cleavage, and retro-hydroalumination.

Regio-controlled synthesis of multisubstituted benzenes is one of the central fields in synthetic organic chemistry because polysubstituted arenes are ubiquitous units in functional molecular materials and biologically active compounds.1,2 Sequential substitution on existing benzene rings is a conventional way of dealing with this subject. However, this strategy is often accompanied by the problems of incomplete regio-selectivity, difficulty in direct introduction of desired substituents, and low efficiency of the linear multistep approach. For the last few decades, much attention has been focused on transition metal-catalyzed cycloaddition of two or more components, which enables a convergent, efficient approach to multisubstituted benzenes. So far $[2 + 2 + 2]$ cycloaddition of two or three kinds of alkynes has been studied intensively.³ Highly selective construction of multisubstituted benzenes by this method appears in recent reports, although the scope is still rather limited.⁴ 1, 3-Enynes and 1,3-dicarbonyl compounds as well as alkynes serve as the components for benzene ring construction.⁵⁻⁷ Yamamoto, Gevorgyan, and co-workers have demonstrated that Pd-catalyzed cyclodimerization of 1,3 enynes and cycloaddition of 1,3-enynes to alkynes form

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	R in R_2 AlH $\left($ equiv $\right)$	temp $({}^{\circ}C)$	time (h)	isolated yield $(\%)$			
entry				2a	3a $(EE:EZ)^b$	1a	
1	i -Bu (1.0)	70	24	43	25(70:30)	θ	
2^c	i -Bu (1.0)	70	24	9	54 (72:28)	Ω	
3	i -Bu (1.0)	30	48	32	32 (47:53)	θ	
4^d	i -Bu (1.0)	110	24	11	e	θ	
5	i -Bu (3.0)	70	24	Ω	e	Ω	
6	i -Bu (0.2)	70	70	42	\boldsymbol{e}	19	
7	i -Bu (0.5)	70	24	46	17(70:30)	Ω	
8^f	i -Bu (0.5)	70	24	50	12(70:30)	Ω	
9 ^g	i -Bu (0.5)	70	24	55	11(60:40)	Ω	
10 ^h	i -Bu (0.5)	70	24	53	19 (62:38)	Ω	
11^i	i -Bu (0.5)	70	24	60	13 (67:33)	Ω	
12^h	Et(0.5)	70	24	67	12 (62:38)	Ω	
13^h	Et(0.5)	70	24	62	15(60:40)	Ω	
14^i	Et(0.5)	70	24	62	\boldsymbol{e}	Ω	

^a Unless otherwise noted, the reaction was carried out with **1a** (0.5 mmol) and R_2 AlH in hexane (1.0 mL). ^bDetermined by ¹H NMR analysis. "With 10 mL of hexane. ^dToluene (1.0 mL) was used instead of hexane. ^e Not isolated due to a complex mixture of products. ^f With 0.5 mL of hexane. ^gWith 0.25 mL of hexane. h ^h Heptane (0.25 mL in entries 10 and 13; 0.5 mL in entry 12) was used instead of hexane. ^{*i*} Without solvent.

multisubstituted benzenes with high regioselectivity. In this context, we herein report that dialkylaluminum hydrides (DIBAL-H and $Et₂AIH$) promote cyclodimerization of silylated 1,3-enynes to unprecedented tetrasubstituted benzenes with complete regio-control. This work provides a rare example of regio-controlled synthesis of multisubstituted benzenes using aluminum reagents.⁸

In the course of our study on synthetic use of vinylsilanes,⁹ we attempted the partial reduction of (E) -1-(trimethylsilyl) dec-3-en-1-yne (1a) with DIBAL-H to synthesize 1-silyl-1, 3-diene 3a (entry 1 in Table 1). However, 3a was obtained in only a low yield. Surprisingly the reaction afforded tetrasubstituted benzene 2a, an unprecedented dimer of 1a, as the major product. No reaction occurred in the absence of DIBAL-H. Our attention was therefore focused on this novel cyclodimerization via skeletal rearrangement.

Optimization of the reaction conditions was initially conducted to improve the yield of 2a. Under the conditions shown in entry 1 of Table 1, several solvents were used instead of hexane. Other less polar solvents (benzene,

toluene, xylene, and 1,2-dichloroethane) allowed the cyclodimerization to 2a in less than 20% yields. Polar solvents such as THF and $Et₂O$ prevented the cyclodimerization completely. The results brought by variation of other conditions are summarized in Table 1. Dilution with a 10-fold amount of hexane was not effective and increased the formation of 3a (entry 2). The cyclodimerization proceeded even at 30 $^{\circ}$ C although the yield of 2a dropped to 32% (entry 3). The reaction at 110 °C resulted in a complex mixture of products (entry 4). An excess amount of DIBAL-H suppressed the dimerization to 2a (entry 5). Longer reaction with 0.2 equiv of DIBAL-H gave 2a in a reasonable yield (entry 6). Use of 0.5 equiv ofDIBAL-H improved the yield of 2a to 46% (entry 7). The reactions using a decreased amount of solvent or under solvent-free conditions were also effective in the cyclodimerization (entries $8-11$). Use of heptane as a slightly less volatile solvent gave a similar result (entry 10). Diethylaluminum hydride (Et₂AlH), a sterically less hindered aluminum hydride, was also used instead of DIBAL-H (entries $11-13$). As a result, the $Et₂AIH-promoted reaction in heptane achieved$ the best result (entry 12).

We next examined the cyclodimerization of 1,3-enynes 1 bearing a different silyl group (Table 2). The DIBAL-Hpromoted reactions of enynes $1b(S_i = SiMe_2Ph)$, $1c(S_i =$ SiMe₂Bn), and 1d ($Si = SiMePh₂$) gave the corresponding tetrasubstituted benzenes in $47-53\%$ isolated yields (entries 1, 3, and 4). Use of $Et₂AIH$ slightly improved the yield of 2b (entry 2). Enyne 1e, bearing a more sterically demanding silyl group ($Si = SiPh_3$), did not undergo the cyclodimerization, and diene 3e and oligomeric compounds were formed (entry 5). Thus, the introduction of a TMS group was found to be more effective in the benzene ring formation than that of other silyl groups.

Table 2. Effects of Silyl Groups^{a}

entry	Si	enyne	product	yield $(\%)$
	PhMe ₂ Si	1b	2 _b	47
2 ^b	PhMe ₂ Si	1b	2 _b	51
3	BnMe ₂ Si	1c	2c	53
4	Ph ₂ MeSi	1d	2d	48
5	Ph_3Si	1e	2e	0^c

 a ^aThe reaction was carried out with 1 (0.5 mmol) and DIBAL-H (0.25 mmol) at 70 °C for 24 h without solvent. \bar{b} The reaction was carried out with 1 (0.5 mmol) and Et_2AlH (0.25 mmol) in heptane (0.5 mL) at 70 °C for 24 h. \degree Diene 3e was obtained in 31% yield along with oligomeric compounds.

To elucidate the scope and limitations of the present cyclodimerization, the reaction of some 1,3-enynes bearing a TMS group was carried out with $DIBAL-H$ and $Et₂AIH$ (Table 3). 4-Cyclohexyl-1-(trimethylsilyl)but-3-en-1-yne (1f) was converted into the corresponding tetrasubstituted

^{(8) (}a) Wilke, G.; Müller, H. Justus Liebigs Ann. Chem. 1960, 629, 222. (b) Eisch, J. J.; Kaska, W. C. J. Am. Chem. Soc. 1966, 88, 2213.

⁽⁹⁾ Miura, K.; Inoue, G.; Sasagawa, H.; Kinoshita, H.; Ichikawa, J; Hosomi, A Org. Lett. 2009, 11, 5066.

benzene in relatively good yields (entry 1). The cyclodimerization of 1g and its Z-isomer 1h proceeded with retention of the alkene geometry (entries 2 and 3). The Z-geometry slightly decelerated the cyclodimerization. Arylated enynes 1i-1 and 1n also underwent the cyclodimerization to give stilbene derivatives in moderate yields (entries $4-7$ and 9).¹⁰ In contrast, enyne 1m was hardly dimerized to 2m (entry 8). The unsuccessful result is likely due to the steric factor arising from the ortho substitution. No reaction occurred with 1,3 enyne 1o, and 1o was recovered quantitatively (entry 10). The ether functionality would make DIBAL-H inactive by coordination to inhibit any reaction. The reactions of 1p (entry 11), 1q, and 1r resulted in complex mixtures of unidentified products. Thus the substituent at the terminal sp^2 -carbon of 1,3-enynes is necessary for benzene ring formation, while the substituent at the internal sp²-carbon prevents it.¹¹ The structure of 2f was clearly determined by X-ray structure analysis.12

Table 3. Scope and Limitations of Cyclodimerization^{a}

R^1 R^2	SiMe ₃ 1f-p	$R2$ Al-H 70 °C, 24 h		$(R^2) R^1$	SiMe ₃ R ² Me ₃ Si 2f-p	R^1		
	enynes 1					isolated yield (%)		
entry	\mathbb{R}^1	R^2		$\overline{2}$	$R = i-Bu^b$	$R = Et^c$		
1	Cy	Н	1f	2f	60	63		
$\overline{2}$	Н	Me	1g	2g	35	51		
3	Me	Η	1 _h	2 _h	56	66		
$\overline{4}$	Ph	Η	1i	2i	44	55		
5	$4-F-C6H4$	Η	1j	2j	40	59		
6	$4 - \text{Cl}-\text{C}_6\text{H}_4$	Η	1k	2k	43	50		
7	4 -Me-C ₆ H ₄	Η	11	21	48	51		
8	2 -Me-C ₆ H ₄	Η	1 _m	2m	trace			
9	$4-Ph-C6H4$	Η	1n	2n	37	44		
10	$4-MeO-C6H4$	Η	1 _o	2σ	Ω			
11	Н	Н	1 _p	$\mathbf{2}\mathbf{p}$	CM^d			

^{*a*} All reactions were carried out with 1 (0.5 mmol) and R_2 Al–H (0.25) mmol) at 70 °C for 24 h. ^bThe reactions were carried out without solvent (entries 1–3, and 11) or in heptane (0.125 mL in entries $4-10$). ^cThe reactions were carried out in heptane $(0.5 \text{ mL} \cdot \text{in} \cdot 1)$ in entries $1-3$; $0.25 \text{ mL} \cdot \text{in}$ entries 4–7, and 9). d CM: a complex mixture of products.

A plausible mechanism for the R_2 AlH-promoted cyclodimerization of silylated 1,3-enynes is shown in Scheme 1.

(12) See the Supporting Information.

First, hydroalumination of the alkynylsilane part of 1 takes place with high regioselectivity (step 1).¹³ The resultant dienylaluminum (Z)-4 is isomerized to (E) -4 (step 2).¹⁴ Intermolecular carboalumination of 1 with (E) -4 forms tetraenylaluminum 5 (step 3), which is cyclized to cyclopentadiene 6 by intramolecular carboalumination (step 4).15 Further carboalumination converts 6 into bicyclic intermediate 7 (step 5). The ring-opening reaction $(C-C)$ bond cleavage) of cyclopropylcarbinyl aluminum 7 yields cyclohexadienylaluminum $\overline{8}$ (step 6).¹⁶ Then the retrohydroalumination of 8 provides 2 with release of R_2A H (step 7).^{17,18} The improved results with Et₂AlH is attributable to the sterically less congested character of the $Et₂Al$ groups, which allows easy access of $C-C$ unsaturated bonds to the reactive sites $(AI-H$ and $AI-C$ bonds) of $Et₂AIH$ and $Et₂AI-based intermediates.$ The reduced bulkiness would also promote the isomerization of (Z) -4 because it minimizes the steric interaction between Z-configurated dialkylaluminum and alkenyl groups in (E) -4.

Scheme 1. Plausible Mechanism

The reaction of enyne 1s, bearing a triethylsilyl group, afforded tetraene 9 and cyclopentadiene 10 with 2s and 3s (Scheme 2). Since the conversion of 1s into 9 can be

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(17) For retro-hydroalumination to regenerate DIBAL-H, see: Ziegler, H.; Martin, H.; Krupp, F. Justus Liebigs Ann. Chem. 1960, 629, 14.

(18) Wilson and Chum reported the DIBAL-H-catalyzed cyclization of α , ω -dienes. The proposed mechanism involves hydroalumination of one C-C double bond, intramolecular carboalumination of another C-C double bond, and retro-hydroalumination with regeneration of DIBAL-H. Chum, P. W.; Wilson, S. E. Tetrahedron Lett. 1976, 16, 1257.

⁽¹⁰⁾ The DIBAL-H-promoted reactions of 1i and 1j without solvent gave 2i and 2j in 40% and 33% yields, respectively.

⁽¹¹⁾ A silyl group is crucial for this transformation. Although the reaction of (E) -1,4-diphenylbut-1-en-3-yne was also carried out, no desired product was formed. (1E, 3E)-1,4-Diphenylbut-1,3-diene and its 1Z-isomer were obtained in 11% total yield (64:36).

explained by intermolecular carboalumination of 1s with dienylaluminum (Z)-4s ($Si = SiEt_3$, $R^1 = Ph$), it partly justifies the validity of the proposed intermolecular carboalumination from (E) -4 to tetraenylaluminum 5 (step 3 in Scheme 1). The formation of 10 is indicative of the presence of intermediate 6s ($Si = SiEt_3$, $R^1 = Ph$). The path from 6s to 10 can be rationalized by hydrolysis of the $C-$ Al bond followed by alkene migration. In the reaction of 1s, the bulky triethylsilyl group probably inhibits intramolecular carboalumination of 6s (step 5 in Scheme 1) to cause the formation of 10.

Scheme 2. Cyclodimerization of Enyne 1s

To ascertain the last retro-hydroalumination step (step 7 in Scheme 1), the reaction mixture from 1a was quenched with D_2O (Scheme 3). ¹H NMR analysis of the product 2a revealed that no D-incorporation occurred. Additionally the cyclodimerization of D-labeled 1,3-enyne 1t afforded 2t without erosion of D-content. These results clearly rule out the possibility of hydrolysis of 8 followed by aerobic aromatization to 2. The retro-hydroalumination step Scheme 3. Deuterium-Labeling Experiments

implies that a catalytic amount of $R₂A$ lH can promote the cyclodimerization of 1. In practice, however, the catalytic use is not feasible for the cyclodimerization of 1 because of many side reactions caused by R_2A IH.

In summary, we have found a R_2A IH-promoted cyclodimerization of silylated 1,3-enynes taking place via skeletal rearrangement. This novel reaction proceeds with complete regioselection to provide rapid access to unprecedented tetrasubstituted benzenes without using transition metal catalysis. This work has revealed that alkenylaluminum reagents have a high potential for regioselective construction of multisubstituted benzenes.¹⁹ Several experimental results suggest an intriguing mechanism for the cyclodimerization. At present, the R_2A IH-promoted crosscoupling between two different 1,3-enynes is under investigation in this laboratory.

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Supporting Information Available. Detailed experimental procedures and characterization data for all new compounds. X-ray crystallographic data for 2f. This material is available free of charge via the Internet at http://pubs.acs.org.

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